

Expert Opinion

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Recent developments in nanoparticle-based drug delivery and targeting systems with emphasis on protein-based nanoparticles

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Background: Drug delivery systems with nm dimensions (nanoparticles [NPs]) are attracting increasing attention because they can sequester drugs in systemic circulation, prevent non-specific biodistribution, and target to specific tissues. **Objective:** We reviewed the recent literature pertinent to NP-based drug delivery, primarily emphasizing NPs fabricated from proteins. **Methods:** A summary of common NP fabrication techniques is provided along with the range of sizes and functional properties obtained. The NP properties critical for injectable drug delivery are reviewed, as well as the attempts to design 'tissue-specific' NPs. **Results/conclusions:** It has been possible to design > 100 nm NPs from different biomaterials, and further understanding of *in vivo* stability and interactions with physiologic systems will lead to improved drug delivery systems.

Keywords: drug delivery, nanoparticles, nanoparticle fabrication, particle size, phagocytosis, surface modification, tissue targeting

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1. Overview of nanoparticulate drug carriers

Injectable colloidal systems (hereon referred to as nanoparticles [NPs]) hold promise for systemic administration of different therapeutic agents, including conventional pharmacologic agents, disease-modifying bioactive factors (e.g., interleukins) and protein growth factors, as well as nucleic acid-based agents intended for gene-based therapies. Nanoparticulate drug delivery systems in the form of solid spheres, micellar emulsions and liposomes (this review does not focus on micellar and liposomal systems and the reader is referred to other reviews on these topics) have been used in the past 20 years as injectable formulations for systemic and tissue-specific delivery of drugs. Several types of biodegradable macromolecules have served as the foundation of drug carriers (Table 1). Conventional synthetic polyesters, whose prototypical examples include poly(lactic acid) (PLA) and poly(lactic-co-glycolic acid) (PLGA), have been extensively used: i) their long history of clinical use brings comfort to their tolerability in the body; ii) their well-understood degradation pattern allows better engineering of drug-release kinetics; and iii) established processing methodologies enable a relatively 'homogenous' product whose performance is better predicted and controlled.

Proteins, primarily members of the albumin family, have served as an alternative biomaterial for carriers, facilitated by the obvious compatibility of this material in physiologic systems. Administration of albumin in the body should not raise any adverse effects as long as the source of the protein is free of transmissible diseases.

Table 1. A representative summary of biomaterials used for the preparation of nanoparticulate delivery systems and drugs used for delivery.

Biomaterial	Drugs	Particle size (nm)	Author [ref.]
Polyester			
Poly(lactic acid)	Protein C	200 – 250	Zambaux [75]
Poly(lactic-co-glycolic acid)	Estrogen	40 – 60	Choi [54]
Poly(ϵ -caprolactone)	Tamoxifen	250 – 300	Chawla [6]
Protein			
Human serum albumin	Doxorubicin	150 – 500	Dreis [29]
Bovine serum albumin	IFN- γ	300 – 340	Segura [26]
β -Lactoglobulin	–	60 – 130	Ko [21]
Protamine	Oligonucleotides	80 – 200	Lochmann [89]
Gelatin	Clodronate	300 – 500	Li [67]
Gliadin	α -Tocopherol/benzalkonium chloride/linalool and linalyl acetate	900 – 950	Duclairoir [90]
Legumin	–	250 – 300	Irache [91]
Polysaccharide			
Chitosan	Ovalbumin	350	Amidi [43]
Polycyanoacrylate			
Poly(isobutylcyanoacrylate), Poly(isohexylcyanoacrylate)	Oligonucleotides	N/A	Fattal [92]
Poly(butylcyanoacrylate)	Doxorubicin	178	Reddy [93]
Poly(hexadecylcyanoacrylate)	–	150	Peracchia [94]
Lipids			
Trimyristin	Paclitaxel	200 – 250	Lee [95]
Glycerol monostearin	Mifepristone	27	Yuan [96]
Glyceryl behenate	Vitamin a	300 – 500	Jenning [97]
Polyelectrolyte			
Polyethylenimine	Ciclosporin	100 – 300	Cheng [98]
Poly(styrene sulfonate)/poly(allylamine.HCl)	Chymotrypsin	100 – 300	Balabushevitch [41]
Poly(L-maleic acid)/Chitosan	Insulin	100 – 250	Fan [42]

Two albumin-based particulate formulations – AlburnexTM as an ultrasound contrast agent and AbraxaneTM as a carrier of anticancer drug paclitaxel [1] – have been accepted for use in humans. Some plant proteins, such as legumin and gliadin, have also been attempted in the development of NP-based drug delivery systems. Due to the unique structure of proteins with multifunctional moieties and/or domains, it might be possible to tailor unique carrier-drug combinations when proteins are used as carriers. For example, albumin – with its hydrophobic core – displays a high affinity for hydrophobic drugs and such combinations may form exceptionally slow-release formulations under physiologic situations. As the primary structure of proteins contains -NH₂, -COOH, -SH and other functional

groups, protein NPs can offer various possibilities for covalent as well as non-covalent (electrostatic) modification of particulate surfaces. The latter is critical for the design of tissue-specific carriers or when physiologic responses to carriers need to be controlled. This is an advantage over polymeric biomaterials, which are rather homogenous and typically provide only a single functional group for modifications. Multifunctional polymers from copolymerization strategies might need to be alternatively used to accommodate various modifications for improved drug delivery.

In addition to proteins and polyesters, other macromolecules used to formulate NP carriers include polysaccharides, synthetic degradable acrylates (e.g., polycyanoacrylate)

and natural/synthetic polyelectrolytes (e.g., chitosan and polyethylenimine [PEI]). The biologic fate of the latter macromolecules is less understood and may pose a challenge for their clinical development. Another attractive idea to achieve controlled drug delivery and release is the use of solid lipid NPs (SLNs), which are primarily derived from middle-chain triglycerides (e.g., trimyristin) or partial glycerides (e.g., Imvitor® 900 and Compritol® 888 ATO) as lipid phase, with other ingredients including emulsifier and water [2].

This review summarizes the recent developments in NP-based drug delivery systems for systemic administration of therapeutic agents, with emphasis on the NPs fabricated from PLA (or PLGA) and albumin, which are two of the most widely used families of macromolecules. The methods for formulating NP carriers, the surface modification of the NPs and the physicochemical properties in the design of drug delivery systems are evaluated; and finally the functionalization of specific tissue-targeting delivery is discussed in detail.

2. Principal preparation methods for nanoparticles

Several distinct methods developed for the preparation of NPs include emulsification solvent extraction, 'salting out' or precipitation, spray drying, solvent displacement, and coacervation and complex formation. Significant overlap exists among some of these methods and it is common to combine more than one process to obtain NPs tailored for specific applications. Each method usually needs to be optimized for entrapment of a drug of interest at high efficiency and retention of the desired pharmacologic activity (especially important for protein therapeutics). An additional requirement for process success is the control over final NP size, usually the diameter. A survey of the literature indicated that most reported NPs are between 100 and 1000 nm. Figure 1 summarizes typical sizes reported for NPs prepared from polyesters, proteins and chitosan as representative biomaterials. Although some processes are able to yield sub-100-nm sizes, this is rare. Three of the more promising methods are emulsion/solvent evaporation, coacervation or desolvation, and polyelectrolyte complexation; Figure 2 schematically outlines the principles underlying these methods.

2.1 Emulsion/solvent extraction process

The emulsification and solvent extraction process is a widely used method for polymeric NPs. It is possible to adopt this technique for protein NPs, although particle sizes are generally larger as compared to the coacervation process described in Section 2.2 [3]. In its simplest form, a polymer solution (in organic solvent [O]) or protein solution (in aqueous buffer [W]) is dispersed in an appropriate non-solvent to form O/W or W/O emulsions, after which

the solvents/non-solvents are removed to form the NPs (Figure 2A). Use of certain organic solvents, such as ethylacetate and chloroform, and surface-active agents, such as poly(vinyl alcohol) (PVA) and polysorbate-80, is not desirable because they may alter the bioactivity of protein therapeutics [4,5]. They may additionally elicit undesired reactions against NPs and need to be reduced to insignificant levels. Surfactant are critical to control particle size: surfactants such as PluronicTM stabilize the formation of smaller fluid particles in emulsions that lead to smaller solid particles [6]. Such a stabilization results from an interaction of the surfactants with the NP matrix. Surfactants are likely to influence drug-matrix interactions and possibly the release rate of drugs in the physiologic milieu [6]. It should be noted that it was possible to prepare PLGA NPs without surfactants [7]. The W/O method was recently used to prepare surface-functionalized NPs after the addition of the desired molecules into the aqueous phase [8]. The W/O/W double-emulsion method can particularly be used for the encapsulation of proteins and hydrophilic drugs [9]; the primary W/O emulsion is introduced into a second water phase to form a double emulsion by using a surfactant for emulsion stabilization. The organic solvent O is then removed and the NPs are finally retained in the aqueous medium.

For PLGA NPs, polymer concentration and solvent composition were the primary factors affecting size [10]. Encapsulation of conventional drugs (e.g., doxorubicin) could be optimized by controlling the ionization state of the drugs [11]. Proteins can be also encapsulated in PLGA NPs with high efficiency (e.g., > 50% [12] or 30 – 50% [13]). For protein NPs, protein concentration and relative volume of solvent:non-solvent during emulsification were the critical parameters [3]. Bovine serum albumin (BSA) NPs with sizes in the range of 100 – 800 nm were reported in a W/O method, dependent on BSA concentration and relative W:O volume ratio [14]. Excellent encapsulation efficiency (> 80%) could be obtained with model macromolecular compounds (e.g., Fluorescein isothiocyanide (FITC)-dextran), partly due to minimal solubility of the compound in the non-aqueous phase. Encapsulation efficiency of conventional drugs is usually lower (e.g., 10 – 60% for tamoxifen) [6]. The bone-inducing morphogen BMP-2 was formulated with PLGA/heparin NPs [15], but protein loading was performed by adsorption post-fabrication, even though prior evidence suggests compatibility of the emulsion process with growth factors such as BMPs [16] and IGFs [17].

2.2 Coacervation process

The coacervation or desolvation process under mild conditions is preferable for the preparation of protein-based NPs and encapsulation of protein drugs. A colloidal system is formed when the solvent used for dissolving the NP matrix is extracted into a non-solvent phase, forming a phase with a colloidal component or coacervate, and a second phase with a solvent/non-solvent mixture (Figure 2B). The particle formation initially proceeds with an increase in size until

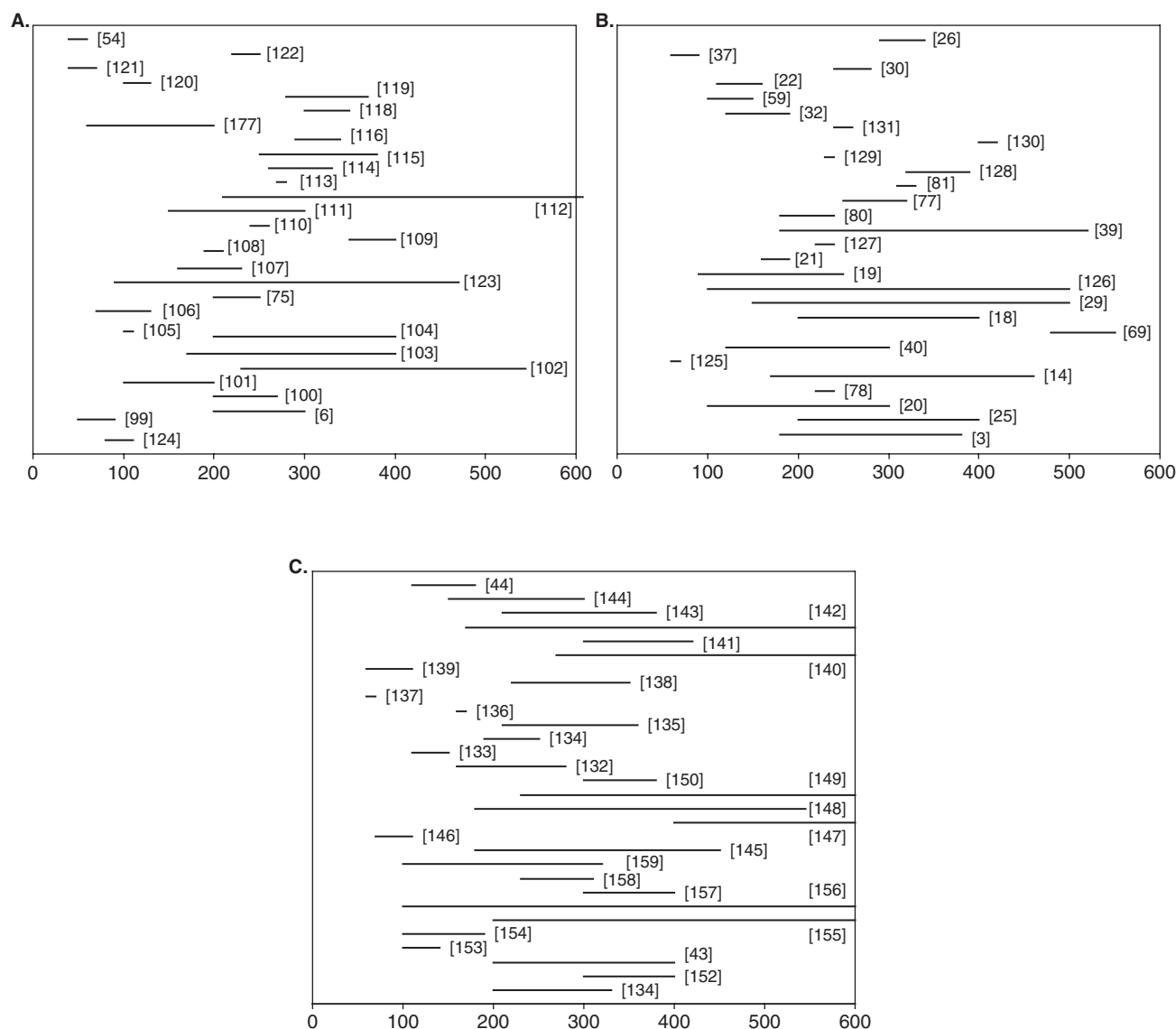


Figure 1. Typical size (nm) ranges of (A) polyester, (B) protein and (C) chitosan nanoparticles reported in the literature. Note that the chosen references were not intended to be exhaustive, but rather representative of common size ranges reported. Chitosan was chosen as a representative polyelectrolyte-based nanoparticle because it has recently received attention for a diverse range of biomedical applications.

a stable size is reached, after which the number of particles gradually increase with increasing desolvation [18]. Lin *et al.* [19] described a pH-controlled coacervation method to prepare ~ 100 nm human serum albumin (HSA) NPs. The particles were prepared by acetone addition to an aqueous HSA solution at pH 7 – 9, and followed by stabilization of the particles by glutaraldehyde (GA) cross-linking. Langer *et al.* [18,20] prepared HSA NPs via desolvation with ethanol; process parameters such as HSA concentration, rate of non-solvent (ethanol) addition, pH of coacervation phase and purification conditions were evaluated and optimized for particle sizes between 100 and 300 nm. The pH prior to

desolvation procedure was identified as the main factor affecting the size, with high pH values leading to smaller particles [20]. This was independently confirmed in our laboratories using ethanol as the non-solvent and BSA as the matrix (Figure 3). The pH and, to a lesser degree, osmolarity were important in maintaining the stability (size) of the particles because conditions leading to net-zero surface charge were more conducive to particle aggregation [20]. Replacing BSA with the smaller β -lactoglobulin (a similar isoelectric point [pI] to BSA) led to a reduction in particle sizes (from 170 to 130 nm with acetone as the non-solvent), presumably due to lower hydrophobicity of

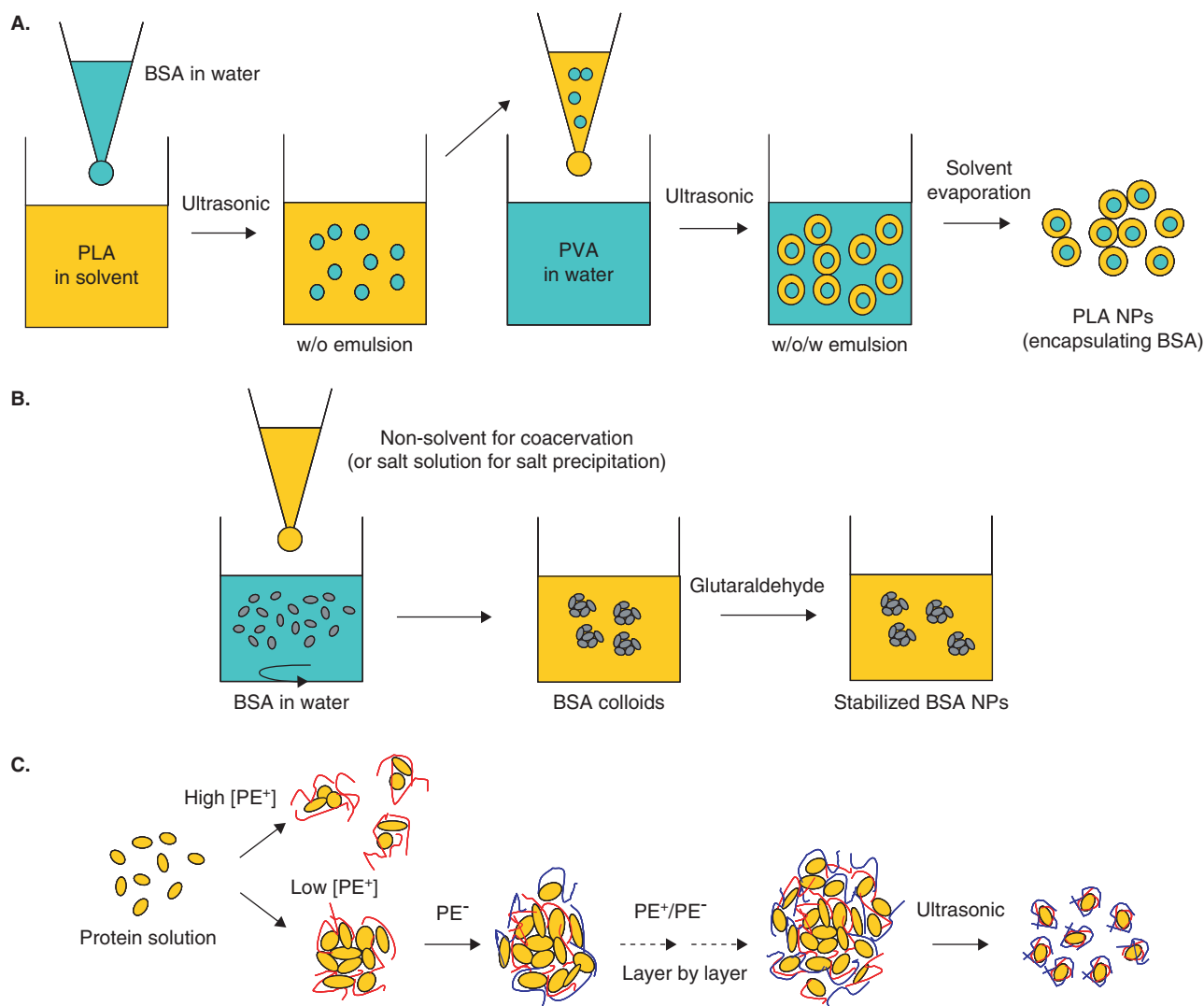


Figure 2. Commonly used methods to prepare nanoparticles: (A) double emulsion; (B) coacervation (and salt precipitation); and (C) polyelectrolyte complexation.

BSA: Bovine serum albumin; o: Oil; PE⁺: Cationic polyelectrolyte; PE⁻: Anionic polyelectrolyte; PLA: Poly(lactic acid); PVA: Poly(vinyl alcohol); w: Water.

the β -lactoglobulin [21]. Denaturation of β -lactoglobulin by heat treatment further reduced the NP sizes to ~ 60 nm and the authors attributed this result to reduced hydrophobic interactions among the extended proteins [21]. Chemically modified proteins may also be used as the NP matrix (e.g., polyethylene glycol-modified BSA [PEG-BSA]), but the use of modified proteins may require changes in the choice of solvents and non-solvents [22].

BSA NPs can achieve very high encapsulation efficiencies for protein drugs. Encapsulation efficiencies in excess of 95% were reported for IFN- γ [23], and Zhang *et al.* observed high entrapment efficiencies ($\sim 90\%$) for BMP-2 in BSA NPs [24]. It was interesting that high entrapment was obtained even with simple adsorption of IFN- γ onto NPs, suggesting a favorable interaction between the NP matrix (BSA) and IFN- γ . This effect is expected

to be drug dependent and, in the case of ganciclovir, entrapment during NP formation was more beneficial than simple adsorption [25].

Despite its known carcinogenicity and, hence, the need for its complete removal before human use, GA has been traditionally used as a cross-linker to stabilize albumin NPs after coacervation. As expected [18], Segura *et al.* [23,26] observed GA cross-linking to reduce the extent of $-NH_2$ groups at the particle surfaces, affecting biodegradability and drug release from the NPs. GA may cross-link the protein amines to BSA matrix, which was also noted for the encapsulation of recombinant IFN- γ , leading to complete abrogation of its bioactivity, while IFN- γ adsorbed on GA-cross-linked NPs retained its bioactivity [23,26]. The GA cross-linking may also be problematic for small molecular drugs, such as doxorubicin [27] and adriamycin [28].

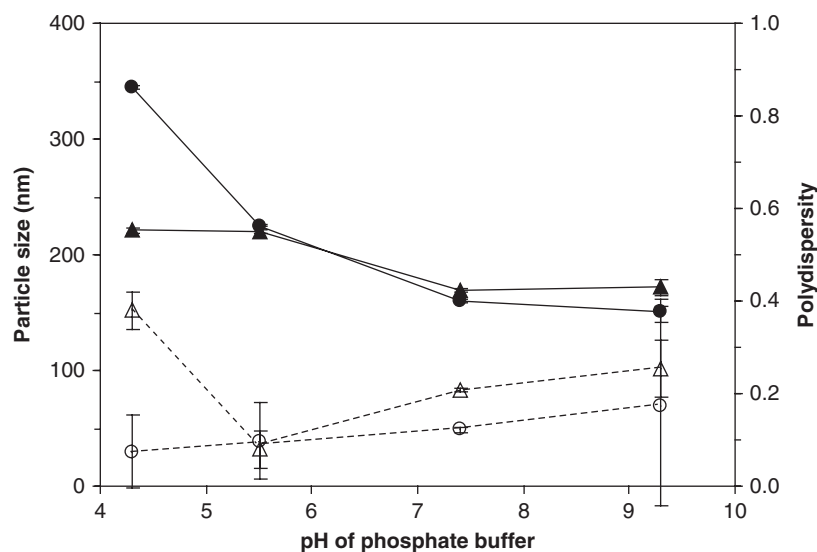


Figure 3. Influence of pH on the size (solid line) and polydispersity index (dashed line) of BSA NPs. NPs were prepared by drop-wise addition of ethanol to a mixture of equal volumes of BSA solution (circle: 10 mg/ml; triangle: 30 mg/ml) and 10 mM phosphate buffer (pH: 4.3, 5.5, 7.4 and 9.3). Size (from dynamic light scattering) and polydispersity data are expressed as mean \pm SD ($n = 3$). Note the gradual reduction of particle size as the pH of the NP preparation solution is increased. BSA: Bovine serum albumin; NP: Nanoparticle; SD: Standard deviation.

An independent study did not indicate any adverse effect of GA on doxorubicin-loaded HSA NPs, although drug release was not assessed in this study [29]. GA cross-linking did not also appear to affect the integrity of an antisense oligonucleotide entrapped in HSA NPs [30], suggesting that GA reaction can be controlled to minimize any adverse effects. An alternative approach is to use high temperature/low moisture cross-linking [18,31], but this may be problematic due to drug deactivation or protein denaturation at high temperature. Macromolecular cross-linkers, such as oxidized dextran, could be alternatively used to replace GA [32] because they are less likely to freely diffuse into the NP matrix and react with the functional groups on the NP surface. A different approach for NP stabilization is to use a non-covalent coating of NP surfaces; BSA NPs could be coated with cationic polymers in aqueous buffers to create a coating that not only improves particle stability, but also controls the release rate of a protein. The feasibility of this approach was demonstrated with BMP-2 entrapped in BSA NPs and coated with PEI [24].

2.3 Polyelectrolyte complexation (complex coacervation)

Macromolecular interactions under aqueous conditions due to electrostatic forces can lead to coacervation complexes (Figure 2C). Formation of insoluble complexes between proteins and oppositely charged polymers will strongly depend on medium pH; use of polyanions for NP formation will require a pH lower than the protein pI, and vice versa for polycations [33]. Cationic polymers, such as

poly-L-lysine (PLL) and PEI, are commonly used for condensing long, string-like DNA molecules into NPs by complexation with anionic oxygen atoms in the phosphodiester backbone of DNA [34]. The driving force for this type of interaction is usually entropic due to the release of the small counterions rather than enthalpic [35]. Comb-type polyelectrolytes were proposed to be particularly suitable for the preparation of water-soluble complexes [35,36] because hydrophilic segments facilitate the aqueous solubility that is essential for systemic application [35]. Serefoglou *et al.* [37] described the preparation of soluble NPs with hydrodynamic radii of 65 – 83 nm through coulombic interaction of BSA with anionic copolymers at a pH lower than the pI of BSA (~ 4.5). These NPs contained, on average, 13 – 14 BSA molecules held together by two polymeric molecules. They were proposed to bear a hydrophobic core of BSA/anionic polymer moieties (~ 20 nm in diameter) and a polymeric corona. In addition to electrostatic interactions, H-bonding or hydrophobic interactions can also drive albumin/polymer complexation [38], the balance between the polymer–polymer and polymer–protein interactions dictating the number of proteins/polymer in a complex. The size of intermolecular complexes was dependent on the number of bound protein/polymer, but its dependence was variable for individual polymers (and possibly interaction mechanisms within the complexes). Rhaese *et al.* [39] prepared HSA-PEI-DNA NPs using complex coacervation induced by charge neutralization as a delivery system for gene therapy. These NPs were much larger in size (300 – 1000 nm) as compared to BSA particles formed at low pH.

Smaller particles with a much broader distribution (30 – 300 nm) were reported by using a combination of protamine, oligonucleotides and albumin; incorporating albumin into such particles was achieved by electrostatic assembly and seemed to significantly influence intracellular distribution of the internalized oligonucleotides [40].

Polyelectrolyte complexation can also be used for coating of protein aggregates by layer-by-layer assembly (Figure 2C [41,42]). This process may provide exquisite control over the structure of the NPs formed, but reported results indicated primary aggregates of 100 – 300 nm in size (similar to BSA NPs) that was further assembled to form microscopic structures as a result of coating. Prevention of larger aggregates in a reproducible way (e.g., with sonication [42]) is paramount to obtaining uniform nano-sized particles. It was possible to control the release rate of proteins from such NPs by controlling the coating layer (i.e., nature of the polymer and the order of coating), providing a convenient method for controlled release [41]. Distinguishing the underlying release mechanism (i.e., protein diffusion versus NP disintegration in a destabilizing environment) will be important for any therapeutic effect obtained.

NP formation by complexation is particularly suited for naturally occurring polymers such as alginate and chitosan. Chitosan offers a unique biologic matrix because it is mucoadhesive and acts as a good substrate for cellular binding. In addition to water-soluble deacetylated chitosan, successful NPs were prepared with *N*-trimethylated (200 – 400 nm [43]) and *N*-(2-hydroxy)-propyl-3-trimethylated chitosan (110 – 180 nm [44]) using tripolyphosphate as the counter polyion. Model proteins (albumin and ovalbumin) were encapsulated with ≤ 80 – 100% efficiency under optimized conditions. As much as 50% of NPs could be made with the protein fraction, indicating the possibility of obtaining protein NPs as a result of polyelectrolyte-mediated aggregation.

2.4 Salt precipitation

A simple approach to the preparation of protein NPs is simply ‘salting out’ a protein solution to form protein coacervates. The simplicity of this approach is attractive, provided that the precipitation process does not change the bioactivity or conformational structure of the protein. For example, insulin particles were prepared in this way by exposing the protein solution to high NaCl concentration (> 0.5 M) at low pH (< 2.0) [42]. The particle sizes were 100 – 1000 nm depending on the pH and appeared to display a large heterogeneity compared to other NP fabrication methods. Coating of such particles with polyelectrolytes can be easily achieved and this helps to stabilize the NPs, as well as to facilitate control under certain conditions [42]. The latter will depend on the pI of encapsulated proteins and the net charge of coating polyelectrolyte(s).

3. Surface modification of nanoparticles with PEG

Considerable effort has been spent to obtain ‘stealth’ NPs with reduced reticuloendothelial system (RES) uptake and prolonged circulation time. One widely used method is surface adsorption or grafting with PEG to create a hydrophilic barrier [45–47], which can block electrostatic and hydrophobic interactions responsible for opsonization. PEG-functionalized PLGA NPs were prepared from PLA-PEG and PLGA-PEG copolymers [48–51]. By using di-block PEG-PLA, PEG-PLGA and PEG-PCL polymers, the influence of PEG molecular weight (MW) and content on the plasma protein adsorption and phagocytosis was investigated [52]. The sterically stabilized PEG surface was found to reduce protein adsorption but not eliminate it. The PEG MW critical for significant reduction of protein adsorption varied between 1.5 and 5 kDa [52,53], and it appeared that the critical MW to prevent protein adsorption might depend on specific protein; that is, a different length of PEG is needed for different proteins. Methoxy-PEG (mPEG)-PLGA block copolymers with different sizes of mPEG block (M_n 550, 750 and 2000 Da) were used for the hydrophilic layer on PLGA NPs [54]. The average particle size was slightly increased with increasing mPEG MW due to extended PEG corona extending into aqueous medium.

Surface coating by adsorption is especially attractive for NP modification because the process can be separately controlled from the fabrication process. Based on a comb polymer of PLL-g-PEG designed for the modification of metal oxide surfaces for protein resistance [55–57], PLL-g-PEG polymers were used to coat PLGA microspheres through electrostatic interactions between cationic PLL backbone and anionic PLGA surface [58]. A drastic decrease in protein adsorption (by two orders of magnitude) was the result. Similarly, protein (HSA) NPs were stabilized with an oxidized dextran-PEG conjugate, ultimately leading to surface grafting of PEG chains [32].

As an alternative to coating or grafting, it is possible to obtain surface-modified NPs by using derivatized proteins in the fabrication process. Lin *et al.* [32] were the first to report HSA NPs prepared from HSA modified with mPEG (HSA-mPEG), poly(amidoamine)-PEG (HSA-PAA-PEG) and poly(thioetheramidoacid)-PEG (HSA-PTAAC-PEG) [22,59]. The existence of a hydrated barrier surrounding the NPs was confirmed and the surface-modified particles showed a reduced plasma protein adsorption. The PEG MW was again important for the prevention of protein adsorption, where a 2-kDa (but not 5-kDa) PEG was most effective. The particles made from PEG-BSA displayed increased stability on salt/pH challenge [22]. The fact that a reduction in protein adsorption was not always proportional to PEG size or the extent of engraftment indicated the possibility of unique interactions with the matrix of the albumin NPs.

Affecting the release rate of entrapped drugs as a result of surface modification is always a concern, but it was possible to achieve NP modification with PEG without affecting the release pattern of drugs [14].

4. Nanoparticle properties critical for drug delivery

Delivery of pharmacologic agents formulated in NPs can be broadly classified into two categories (i.e., those intended for local versus systemic administration). In the case of local administration, the NP formulation is expected to act as a 'depot', where a local concentration of the drug is maintained while preserving the agent in active form, usually in a hostile environment. The NP may also act as a controlled-release formulation, where NP properties precisely control the drug release rate. In the case of systemic administration, the NP formulation is expected to circumvent the initial clearance mechanisms in circulation and preferably localize at the site of action of the drug. Once successfully localized to a desired site, the NP formulation becomes a local depot formulation. Understanding relationships among the physicochemical properties of NPs and their physiologic performance is paramount for successful delivery. Towards this end, a summary of the pertinent literature follows, specifically focusing on NP properties critical for systemic and local delivery.

4.1 Particle size and circulation time

Long-circulating carriers have a better chance of reaching their target and leading to a therapeutic benefit. Larger NPs are more likely to sediment in circulation or filtered in capillary beds and it is generally accepted that the NPs should not exceed 200 nm for this reason [60]. Despite several reports with exquisite control over NP size, the effect of size on circulation time is understudied. Using relatively small BSA NPs (110 – 125 nm), PEG functionalization of NPs was shown to improve the circulation time (systemic mean retention time increased by ~ 1.5-fold) as well as localization in some tissues, in particular the brain [14]. Harashima *et al.* [61] reported the size dependence of opsonization (see Section 4.2 for more details on opsonization process) for liposomes, suggested that smaller carriers led to a reduced adsorption of proteins and opsonins, and accordingly reduced uptake by phagocytic cells. The extent of opsonization decreased with a reduction in size from 800 to 200 nm, and no enhancement of phagocytic uptake was recorded at < 200 nm. Besides systemic clearance, biodistribution of the NPs is also affected by their size. Using PEG-decorated NPs in rabbits, Porter *et al.* [62] showed that particles < 150 nm exhibited an increased localization in bone marrow, while particles ~ 250 nm were mostly sequestered in the spleen and liver, presumably by selective filtration in these organs [60]. In the bone marrow, a membrane of lining cells functions as a marrow-blood

barrier and not all NP localized to bone are actually accessible to bone-resident cells [63]. It is suggested that the vasculature in bone has pores of 80 – 100 nm [54,64] and NPs need to have a hydrodynamic size of < 80 nm to extravasate into bone tissue. However, particles < 100 nm may also pass through fenestrated hepatic sinusoidal endothelium [65] and may get entrapped in the liver, ultimately reducing their availability to the bone-resident cells.

For some applications, it is desirable to engineer NPs that can be taken up by the phagocytic cells. Immunostimulation of macrophages against microbial infections or immunosuppression of macrophages in auto-inflammatory diseases may be achieved with large NPs of 200 – 300 nm [23]. Gelatin NPs containing the cytotoxic drug clodronate, a bisphosphonate drug usually formulated with liposomes [66], were successfully used for the elimination of macrophages [67].

4.2 Nanoparticle opsonization and phagocytosis

A major obstacle for particulate drug carriers is the opsonization and clearance by the RES. The opsonization process is the adsorption and/or activation of proteins capable of interacting with specific receptors on monocytes and various subsets of tissue macrophages, thus promoting particle recognition and uptake [65]. As much as 95% of foreign particles may undergo opsonization and clearance by the RES [45]. Opsonization typically takes place in the blood circulation and can take from a few seconds to many days to complete [68]. The exact mechanism through which this process is activated is not yet fully understood. The small size of the NPs is no impediment for cellular uptake, and sizes and surface characteristics have been repeatedly emphasized as important factors affecting clearance behavior and tissue distribution of particulate carriers [65]. Neutral and hydrophilic particles have a much lower opsonization rate than the charged and hydrophobic particles [69]. The core properties of the NPs, although generally considered shielded from the physiologic contact, may also affect the extent and nature of protein adsorption [52]. PEGylation is the standard approach to reduce protein adsorption to NPs (see below), but some degree of protein adsorption might be beneficial to prevent NP aggregation if the surface properties alone are not sufficient (e.g., low PEG density) to prevent aggregation [32]. Recent studies with poly(isobutylcyanoacrylate) NPs indicated that the particle size (100 – 300 nm) do not influence the extent of complement activation [70]. However, surface chemistry, specifically the choice of polysaccharide coating, was important with chitosan coating giving relatively low complement activation that could be reduced to background levels with high-MW chitosans. Whereas the size of naked poly(cyanoacrylate-co-n-hexadecyl cyanoacrylate) NPs did not influence protein adsorption, PEGylation of such NPs resulted in lower protein adsorption and phagocytic uptake *in vitro* with decreasing particle size [71]. However, the ideal

PEG MW for effective reduction of opsonization was variable in different NP preparations and this was likely to reflect the exact configuration of the PEG on the NP surfaces [72].

With BSA NPs, the negative ζ -potential (-15 to -30 mV) was considered a significant advantage because it can reduce non-specific interaction with cellular surfaces. However, despite an anionic surface, ~ 200 nm HSA NPs were readily internalized by primary human macrophages [20]. Cationic gelatin NPs (300 – 500 nm) were also internalized by the rat macrophage-like cells [67]. The effect of particle size and internalization was not systematically investigated for protein NPs, but having highly anionic and cationic surfaces was equally detrimental for uptake in macrophage-like U937 cells and mouse macrophages [69]. Internalization of cationic NPs was not surprising, but anionic NPs were presumably opsonized for such a stimulated uptake [69]. Unlike *in vitro* results, *in vivo* uptake into the major clearance organ (liver) was not affected by the ζ -potential of these particular NPs (size ~ 500 nm) [69].

4.3 Factors controlling drug release

Mechanism(s) controlling drug release from NPs may include: i) dissolution of drugs from a solid state; ii) displacement of adsorbed drugs from NP matrix; iii) diffusion through NP matrix or surface coating (if any); and iv) degradation of NP matrix or surface coating [46]. For drugs strongly interacting with a NP matrix [6], displacement with buffer components (*in vitro*) or endogenous solutes (*in vivo*) may be a critical factor. With conventional drugs of high diffusivity, such as ganciclovir [25] and clodronate [67], release patterns indicated two types of drug fractions: a weakly and a strongly bound fraction. The former fraction is released following changes in medium composition, where the drug solubility is enhanced or displacing moieties appear in the new medium. A rapid release could be observed without a distinctive 'burst' versus 'sustained-release' pattern [14] and the extent of the initial release is likely to depend on the loading method [25]. NPs prepared from synthetic polyesters, as compared to proteins, might be better engineered for a smaller burst release [73,74], and longer duration of release [12,75,76] by controlling the MW and/or composition of polymers used for NP fabrication [74,75]. Although sustained release is usually desired, a 'burst' release might be sufficient for some indications; IFN- γ released within a 24-h period was sufficient for lasting immunostimulatory activity [23].

Enzymatic degradation is an important mechanism for protein and polyester NPs. It is usually surface restricted [6] and may cause accelerated NP degradation due to a large surface area:volume ratio. Little work has been performed in assessing the degradation of NPs under physiologic conditions. Wartlick *et al.* [77] showed a correlation between the extent of GA cross-linking and rate of enzymatic degradation of HSA NPs *in vitro*. This was also the case

after NP internalization, where HSA particles remained intact when NPs were excessively cross-linked, but the NP cargo was dispersed in cytoplasm for undercrosslinked particles [77]. Because these studies relied on fluorescently labeled molecules and the integrity of the label was not explored, these results need to be considered with caution. BSA and β -lactoglobulin particles also displayed a relatively rapid (half life [$t_{1/2}$] ~ 8 – 15 h) enzymatic degradation *in vitro* [21]. The effect of degradation on pharmacologic activity needs to be probed for individual indications, as well as the correlation between the drug release rate and enzymatic degradation.

5. Designing nanoparticles to target specific tissues

An exciting possibility for NPs could be realized if they can be made 'tissue seeking'. Although particle size and surface characteristics may alter biodistribution *in vivo*, tissue specificity can only arise if NPs display a preferential affinity to a target tissue. To accomplish this goal, NPs are functionalized with tissue-specific molecular ligands. These ligands usually display an affinity to tissue-specific cell surface molecules and they improve the retention of NPs due to enhanced local affinity and slowing of NP loss from the tissues. Alternatively, substrates for tissue-specific transporters have been used for NPs to improve cellular uptake (i.e., cells responsible for barrier functions, such as the brain–blood-barrier [BBB]) into the desired tissues. While natural ligands have been preferred, synthetic ligands with no biologic function have also been used for this purpose. The impetus for the latter ligands is the superior affinity of the ligands to target tissues (e.g., bisphosphonates for bone), or an ability to tailor target affinity and specificity via chemical engineering (e.g., aptamers). A summary of current efforts for tissue-targeting NPs is provided in Table 2 and a select set of applications is discussed below.

5.1 Brain targeting

Targeting NPs to the brain has relied on ligands that display facilitated transport at the BBB. Transferrin is one such ligand and Mishra *et al.* [14] described modification of BSA NPs with maleimide-PEG-*N*-hydroxysuccinimide (MAL-PEG-NHS) so as to functionalize the NP surfaces with transferrin. Only a small fraction of surface PEG groups were coupled to transferrin (3 – 4%) so that an effective PEG coating for improved biodistribution could be achieved by this method. Delivering azidothymidine (AZT) to the brain in transferrin-PEG-HSA NPs was two- to threefold more effective than the PEG-HSA NPs, which was more effective as compared to free AZT or AZT entrapped in unmodified NPs [14]. Liver uptake was also enhanced for transferrin-functionalized NP and these two organs – at times – accounted for ~ 50% of the administered dose [14]. NPs functionalized with apolipoprotein E were also prepared

Table 2. A summary of ligands used for NP derivatization for targeting NPs to specific tissues.

Ligand	NP derivatized	Target cells/tissue	Evaluation	Ref.
Folate	BSA	Cancerous cells	<i>In vitro</i>	[125]
Folate	PLGA	Cancerous cells	<i>In vitro</i>	[160]
Aptamer against PSMA	PLGA	Tumor	<i>In vivo</i>	[83,84]
Wheat germ agglutinin	PLGA	Tumor	<i>In vivo</i>	[161]
HER2	HSA	Cancerous cells	<i>In vitro</i>	[80,81]
HER2	PLA or PLGA	Cancerous cells	<i>In vitro</i>	[76,82]
GFdTGFLS-glucose	PLGA	Brain	<i>In vivo</i>	[162]
Apolipoprotein E	HSA	Brain	<i>In vivo</i>	[78,79]
Synthetic opioid peptides	PLGA	Brain	<i>In vivo</i>	[163]
Transferrin	HSA	Brain	<i>In vivo</i>	[14]
Bisphosphonate	Liposome	Bone	<i>In vitro</i>	[87]
Bisphosphonate	PLGA	Bone	<i>In vitro</i>	[54]
Cyclic RGD	PLGA	Endothelial cells	<i>In vitro</i>	[164]
Linear GRGDS	PLGA/PLGA-PEG/PCL-PEG	M cells	<i>In vitro</i>	[13]
Tetanus toxin C	PLGA-PEG	Neuronal cells	<i>In vitro</i>	[50]

GRGDS: ; HSA: Human serum albumin; NP: Nanoparticle; PCL: ; PEG: Polyethylene glycol; PLA: Poly(lactic acid); PLGA: Poly(lactic-co-glycolic acid); PSMA: Prostate specific membrane antigen.

for brain targeting by direct coupling with the MAL-PEG-NHS linker [78,79] or indirect coupling via an avidin/biotin bridge [79]. A desired pharmacologic effect was observed only when the drug of interest, loperamide, which is impermeable to the BBB, was formulated in apolipoprotein E-decorated NPs [78,79]. The extent of NP targeting as a result of apolipoprotein E coupling remains to be quantified. By using a variety of isoforms, the mechanism of NP transport appeared to be specific for apolipoprotein E and it relied on the transcytosis pathway undertaken by lipoprotein particles.

5.2 Tumor targeting

Surface modification of BSA [80,81], gelatin [81], PLA [82] and PLGA NPs [76] with anti-human epidermal growth factor receptor 2 (anti-HER2) antibodies was attempted to target NPs to breast cancers. The uptake of antibody-coupled NPs was facilitated by HER2-expressing tumor cell lines *in vitro*. However, on prolonged incubation, even non-specific NPs (e.g., PEG-HSA) were internalized [80]. This was unexpected because: i) the NPs displayed a ζ -potential of ~ -40 mV (repulsive to anionic cell surfaces); and ii) PEG molecules should have reduced non-specific cell surface interactions. Sedimentation might be one mechanism for depositing NPs on the cell surface, but how this translates to the *in vivo* situation remains to be seen. PLA NPs functionalized with a control (non-internalizing) antibody gave little uptake as compared to HER2-mediated uptake [82], suggesting that the internalization was specific and not

due to chemical derivatization on the surface. *In vivo* studies on these NPs remain to be reported. On the other hand, PLGA NPs functionalized with aptamers against prostate-specific membrane antigen yielded approximately fourfold improved targeting of the carriers in a xenograft prostate graft model [83] and significantly improved the survival in a xenograft model as compared to non-targeted chemotherapeutic delivery [84].

5.3 Bone targeting

The structural qualities of bone – especially the presence of hydroxyapatite (HA) – and specific affinity of certain molecules to this mineral provide a unique opportunity to target drugs to bone tissue. Molecules with diverse structural features were reported with bone affinity [85,86]. Unlike natural molecules used for targeting to other tissues, bone-targeting ligands are generally synthetic entities, among which bisphosphonates (BPs) play a prominent role. The BPs are a class of synthetic compounds structurally related to pyrophosphates, but they feature a geminal diphosphonate (P-C-P) instead of the hydrolyzable P-O-P bond of pyrophosphate. BPs are highly water-soluble, acidic compounds at physiologic pH, and remain bound to bone mineral after systemic administration. BPs have been traditionally conjugated to therapeutic molecules for bone targeting, but using NPs functionalized with BPs might be a superior alternative; bone targeting can be achieved without modifying the pharmacologic agent *per se*, an important consideration particularly for bioactive protein therapeutics.

The design of bone-seeking PLGA NPs (micelles) was reported by using a PLGA polymer grafted with alendronate [54]. The NPs were 40 – 60 nm in size (a suitable size for vascular penetration at bones) and were able to entrap and release a hydrophobic drug (estrogen) from its PLGA core. Although the alendronate-incorporated NPs displayed hydroxyapatite affinity, it was not clear whether the binding was due to alendronate *per se*, because binding of unfunctionalized NPs were not reported. BPs were also incorporated into liposomes by using a lipophilic BP derivative, cholesteryl-trisoxymethylene-bisphosphonic acid (CHOL-TOE-BP) [87]. Such liposomes were 100 – 135 nm in size, and displayed an *in vitro* hydroxyapatite affinity that was dependent on the extent of BP incorporated in liposomes. The pharmacokinetics and tissue distribution of the BP-conjugated NPs and liposomes remain to be probed to explore their full potential to act as bone-seeking NPs.

Future studies in bone targeting are likely to be accelerated because the basic foundation of this approach is being steadily established in independent laboratories. Synthetic polymers with BP moieties have been synthesized and their superior bone affinity demonstrated [64]. Such polymers can be easily fabricated into bone-seeking NP formulations. Our extensive experience on protein derivatization with BPs could yield bone-seeking protein NPs if the derivatized proteins can be fabricated into NPs. Cationic polymers, such as PLL and PEI, were recently reported to display a hydroxyapatite affinity that was equivalent to BP-mediated affinity [88]. Such polymers can be used in coating NPs for bone targeting.

6. Conclusions

Pharmacologic agents entrapped in NPs are attracting increasing attention to overcome delivery challenges posed *in vivo*. Well-established approaches to fabricate NPs are continually optimized to improve the encapsulation efficiency while preserving the bioactivity of therapeutic agents. New methods to formulate NPs are being proposed; molecular assembly along with the precise control of biomaterial structures are leading the way in this effort. While the size of NPs from several fabrication processes is now routinely controlled, understanding the impact of size on various facets of drug delivery – such as phagocytosis, circulation time and drug release kinetics – remains understudied. Several avenues have been taken to functionalize the surface of NPs with desired ligands. These include NP fabrication from modified biomaterials, coating NPs with designed polymers, surface-grafting post-fabrication and using non-covalent ‘affinity’ bridges to place ligands on NP surfaces. These efforts are establishing the foundation of tissue-specific drug delivery by NPs. Therapeutic applications of tissue targeting are expected to broaden and more efforts will be needed to better

understand the *in situ* fate of NPs, as well as the relationship among NP degradation, drug release and observed therapeutic effects.

7. Expert opinion

NPs fabricated from native proteins are attractive due to the existence of natural mechanisms for metabolizing such particles. Once internalized, enzymatic degradation of such particles is expected to be rapid. *In vitro* studies indicated rapid particle degradation (within 24 h), but it is not possible to readily extrapolate these results to the *in vivo* situation mainly due to an arbitrary choice of the components of the *in vitro* system (i.e., choice of enzymes and concentrations, biologic milieu and so on). Rate of *in vivo* degradation of NPs remains to be fully explored and this will shed light on drug pharmacokinetics delivered with NPs [74]. One important aspect related to protein NPs is the possibility of inducing an autoimmune response, whereby inflammatory or immune response against synthetically produced NPs may turn against the host. This is especially important for application where NPs need to be administered repeatedly. *In vivo* residence time of the NPs is expected to be one important determinant of such a reaction, because rapidly metabolizing systems are less likely to elicit such a response. Clinical use of recombinant proteins had raised the same issue initially, but long history with such proteins has clarified the acceptable benefit–risk relationships for such drugs. However, two additional concerns in the case of NPs are: i) solidified/bulk nature of administered proteins, unlike recombinant proteins that are administered as soluble proteins; and ii) possibility of significant denaturation during NP fabrication. It is difficult to predict the chances of NPs raising an autoimmune reaction based on animal studies, but should it be a limiting concern, synthetic polymers are expected to provide an alternative in this regard. Our survey of the literature indicates that particles with desired sizes could be made from synthetic polymers, proteins and natural polymers (Figure 1), so that a wide variety of biomaterials is available for a given size of NP preparation.

From a technologic perspective, a better understanding of the factors controlling release rate as well as the ways to prolong the release will be greatly beneficial. It is recognized that ‘burst’ release could be sufficient for some applications in preclinical models; however, even for these applications, a sustained release will translate into clinical benefit if the administration frequency could be reduced. Based on published reports from independent laboratories, NPs from synthetic biomaterials appear to provide a longer-release duration, whereas NPs prepared from protein and natural polyelectrolytes provide a large burst-release and shorter sustained-release period. The sustained release, which is usually quantified under *in vitro* conditions, cannot be translated to the *in vivo* situation due to a lack of realistic degradative mechanisms

in vitro. As highlighted earlier (see Section 4.3), determining the physiologic half-life of NPs in various tissues will provide a good measure of the maximal sustained release that can be achieved. Understanding the factors that determine NP stability in various organs is likely to lead to engineering NPs with tailored degradation rates and, hence, sustained release. Very little, if any, work has been performed that probes the internal structure of the NPs. It is likely that significant heterogeneity exists in NPs and that release rates are closely linked to the internal structure. This is particularly relevant to tissue-targeting efforts; *in situ* residence of a

NP coupled with practical release rates/duration will dictate the success of the tissue-targeting efforts.

Declaration of interest

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